

$pK_a$  and  $pK_{aH}$  values of Sulfoximines from






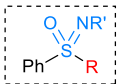

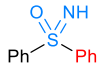
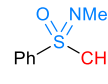
## Focus on Physico-Chemical Properties of Sulfoximines: Acidity, Basicity and Lipophilicity

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**Table 2.**  $pK_a$  and  $pK_{aH}$  values in MeCN of sulfoximines **4a-e**, **5a-5c**

					
<b>4a</b>	<b>4b</b>	<b>4c</b>	<b>4d</b>	<b>4e</b>	
					
	<b>5a</b>	<b>5b</b>	<b>5c</b>		
<b>R</b>	<b>R'</b>	<b>Experimental <math>pK_a^a</math></b>	<b>Computed <math>pK_a^b</math></b>	<b>Experimental <math>pK_{aH}^a</math></b>	
<b>4a</b>	CF <sub>3</sub>	H	<b>26.98 ± 0.12</b>	27.2 ± 1.2	<b>2.08 ± 0.10</b>
<b>4b</b>	CF <sub>2</sub> Br	H	<b>27.0 ± 0.4<sup>c</sup></b>	26.7 ± 1.2	<b>2.59 ± 0.06</b>
<b>4c</b>	C <sub>4</sub> F <sub>9</sub>	H	<b>26.48 ± 0.06</b>	26.3 ± 1.2	<b>1.24 ± 0.07</b>
<b>4d</b>	CF <sub>2</sub> H	H		29.8 ± 1.2	<b>5.28 ± 0.06</b>
<b>4e</b>	CH <sub>2</sub> F	H		31.9 ± 1.2	<b>7.93 ± 0.05</b>
<b>5a</b>	CH <sub>3</sub>	H		34.9 ± 1.2	<b>11.24 ± 0.10</b>
<b>5b</b>	Ph	H		33.1 ± 1.2	<b>9.88 ± 0.05</b>
<b>5c</b>	CH <sub>3</sub>	Me		≥40 <sup>d</sup>	<b>12.32 ± 0.05</b>

<sup>a</sup> The uncertainties were evaluated to safely cover the results from measurements against different reference compounds (see the SI) and we estimate their coverage probability as approximately 80% in the context of the current  $pK_a$  and  $pK_{aH}$  scales; <sup>b</sup> Computationally predicted  $pK_a$  values of sulfoximines in MeCN; Details in the SI (section II.3). <sup>c</sup> When measuring compound **4b** some side-process(es) took place and most measurements were unsuccessful. This measurement was the only successful measurement. The nature of the compound and the sole measurement are the reason for the high uncertainty. <sup>d</sup> See the SI for details.